

Some mechanical properties of pure Ti make it inappropriate for certain industrial or biomedical techniques, so it is necessary to enhance these mechanical properties through alloying.

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n the current work, the microstructure, mechanical properties and machinability of Ti15SnxCu alloys (x = 0 to 2 wt. %) were investigated. The alloy was prepared by the vacuum arc, re-melting and casting route, and then thermally processed by solutionizing at 1,000°C for two hours, followed by water quenching. By controlling the heat-treated condition and Cu content, the highest bending strength and modulus were obtained from three-point bending testing. The increases were mainly caused by the strengthening effects of the formation of crystallites with fine grain size and nano-Ti2Cu precipitate phase. In addition, nano-Ti2Cu precipitation was caused by crack initiation and propagation, which reduced the ductility and improved the machinability of Ti15Sn2Cu alloy.

1 INTRODUCTION

In dentistry, alloys of Ni-Cr, Co-Cr, Ag, and Au are commonly used to manufacture crowns, bridges, castings, inlays, and denture bases. Recently, titanium and titanium alloys have also been used for these applications [1,2]. Due to their low density, excellent biocompatibility, corrosion resistance, and mechanical properties, pure Ti and its alloys are widely used for many biomedical applications today [1-3]. But some mechanical properties of pure Ti make it inappropriate for certain industrial or biomedical techniques, so it is necessary to enhance these mechanical properties through alloying [4,5]. Another, α + β type T64 (Ti6Al4V) is the most widely used because of its better physical and mechanical properties vs. commercially pure titanium (CP-Ti) [6]. However, the release of Al and V ions from the alloy might cause serious problems [7,8]. Therefore, many efforts have been working out to develop non-toxic elements, Ti alloys such as Ti-Cu 9,10], Ti-Sn [11], Ti-Cu-Sn [12,13], Ti-Nb [14,15] and Ti-15Zr-Mo alloys [16]. It is well known that alloying elements alter the transformation temperatures, phases/crystal structure of titanium alloys [4,5,17].

Copper (Cu) is a β stabilizer element. The fusion temperature of the alloy decreases with increases in the amount of copper, which greatly favors casting procedures [18]. The Ti2Cu phase diagram indicates the formation of α -Ti and precipitation of Ti-Cu intermetallic at a copper concentration of 7.0 wt. % [19]. Many authors have focused on the use of Ti-Cu alloys for industrial applications. Kikuchi et al. [18,20] reports that Ti2Cu intermetallics increase the tensile strength and decrease the ductility of Ti-Cu alloys as compared with CP-Ti. Sun et al. [21] showed that Ti2.5Cu alloy was strengthened by the formation of Ti2Cu intermetallic particles after the decomposition of β phase into α -Ti phase and Ti-Cu. Yao et al. [22] reported acicular Ti2Cu particles made greater strengthening contributions than spherical ones for Ti2.5Cu alloy.

Recently, the author confirms the addition of 2 wt% Cu, the microstructure of Ti15SnxCu alloy, displayed a nano-(Ti, Sn)2Cu phase precipitated in α '-Ti(Sn,Cu) with TEM [23]. The 2.5 wt% Cu addition to Ti6AlxV(x = 1.5 and 4) already has been commercially used to nano-Ti₂Cu precipitates [24]. Liu et al. [25] reported that the antibacte-

rial rates of the Ti–Cu alloy against E. coli and S. aureus after 7 h of incubation are 99.01% and 91.66%, respectively, indicating that the copper ions release from the Ti-Cu alloy and contribute to the strong antibacterial property, but the Cu ion did not lead to cell cytotoxicity.

Tin (Sn) is widely used in biomedical Ti alloys. Niinomi M. has shown binary Ti-Sn alloy has been found nontoxic and nonallergic [26]. Therefore, Sn as major alloying element is safe to use with Ti alloy. In addition, Sn, a neutral element, has little influence on the transformation temperature. Sn can react with Ti and Cu in the liquid state, leading to a high nucleation rate and limited grain growth upon solidification [26]. This limited grain size significantly increases the alloy strength [27]. Tsao et al. [12] established that the addition of Sn to Ti7CuXSn (X = 0 - 5 wt. %) alloys can effectively increase the solubility of Cu in the eutectoid microstructure (α-Ti + Ti₂Cu), induce the transformation of an ultrafine α ' martensite microstructure, and increase the volume fraction of the Ti₂Cu phase. Han et al. [28] found increasing the Sn content (1-3 at. %) in Ti alloys can decrease the lamellar spacing in the eutectic matrix to improve the plasticity and strength. Hsu et al. [12] reported alloy hardness is directly proportional to Sn content in Ti-Sn alloy. Ho et al. [29] showed Sn content was significantly related to the machinability of the TixSn (x = 1 to 30 wt. %) alloys. When the Sn content was 20 wt. % or greater, the alloys showed a combination of brittleness and some ductility properties. He et al. [30,31] found that higher Sn content would shift the alloy composition from hypoeutectic to hypereutectic in the Ti-Sn phase diagram, and that only the compositions of Ti-Sn alloy at the hypoeutectic region could promote the formation of a dendrite/ultrafine structure with high strength and distinct plasticity. The Ti-Sn-Cu alloy has a potential application as a dental implant material.

Therefore, the aim of this study was to evaluate the microstructure and mechanical properties of Ti15SnxCu (x = 0 and 2 wt. % Cu) alloy samples obtained by as-casting and heat treatment.

2 EXPERIMENTAL PROCEDURE

2.1 Specimen preparation and microstructure

Ti15SnxCu samples with x = 0 and 2 wt. % were prepared from commercially pure Ti (grade 2), Sn (> 99.99 wt. %) and Cu (> 99.99 wt. %) metals using vacuum arc melting furnace. An arc melting process with a tungsten electrode on a water-cooled copper hearth was used to melt the samples. These samples were prepared in high purity argon atmosphere (> 99.999% pure). The Ti15SnxCu samples were repeatedly melted a total of five times at 1,600-1,630°C for 2-3 min and directly cast in a copper mold. The obtained ingots were solution-treated at 1,000°C in argon atmosphere for 2 hours and then quenched in water to room temperature.

All samples were ground with silicon carbide papers up to 2,000 mesh, polished, and etched (Keller's etchant, 1mL of HF, 2.5mL of HNO $_3$, 1.5mL of HCl and 95mL of H $_2$ O) to reveal the microstructure. The morphologies were studied using optical microscopy (OM)

and scanning electron microscopy (SEM, S-3000H, Hitachi Co.) with a voltage of 20 keV. Energy dispersive spectroscopy (EDS) was used to analyze the composition.

2.2 Mechanical property

Three-point bending tests were conducted on a material testing system at a ramp rate of 1 mm/min. According to ASTM E855, the maximum bending strengths were determined with the following equation [32]:

$$\sigma = 3 \frac{PL}{2bh^2}$$
 Equation 1

Where σ is the bending strength (MPa), P is the load (N), L is the span length (mm), b is the specimen width (mm), and h is the specimen thickness (mm). The mechanical properties were evaluated with a three-point bending test [32,33]. The samples were cut from the central part of the ingots to dimensions of 2.0 (b) mm x 2.0 (h) mm x 20.0 (l) mm. All samples were polished with 1,200-grit SiC paper, and cleaned. The modulus of elasticity in bending was calculated from the load increment and the corresponding deflection increment between two points as far apart as possible on a straight line using the equation [32,33]:

$$E = \frac{L^3 \Delta P}{4bh^3 \Delta \delta}$$
 Equation 2

Where E is the modulus of elasticity in bending (Pa), ΔP is the load i (N), and $\Delta \delta$ is the deflection at midspan. The average bending strength and modulus of elasticity in bending were taken from five tests under each condition.

2.3 Machinability

Machinability was evaluated by the cutting length. The end mill traveled for 30 secs from one edge of the sample at a cutting depth of 0.1 mm. The cutting feed speed of the end mill was 12 mm/min and the cutting speed was 30 m/min. The end milling process are carried out on a CNC vertical milling center using a TiAlN-coated carbide end mill (Kennametal HER 060NN4-E) of diameter 6 mm. No water-soluble cutting fluid was used. Five samples were used to

evaluate the machinability, and the test was performed twice for each kind of sample at each cutting condition. After testing, the milling surfaces and metal chips were investigated by SEM and the average chip metal length for each kind of sample was calculated.

3 RESULTS AND DISCUSSION

3.1 Microstructural characterization

The microstructures of the Ti15SnxCu alloys in the different treatment conditions are shown in Figures 1 and 2. Figure 1a shows the OM image of AS Ti15Sn alloy had the typical Widmanstätten, basket-weave

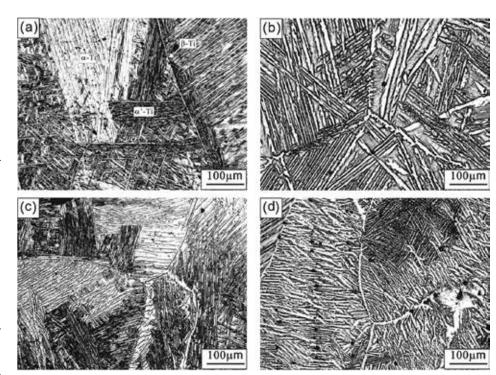


Figure 1: Optical micrographs of the Ti15SnxCu alloys: (a) AS Ti15Sn, (b) HT Ti15Sn, (c) AS Ti15Sn2Cu and (d) HT Ti15Sn2Cu.

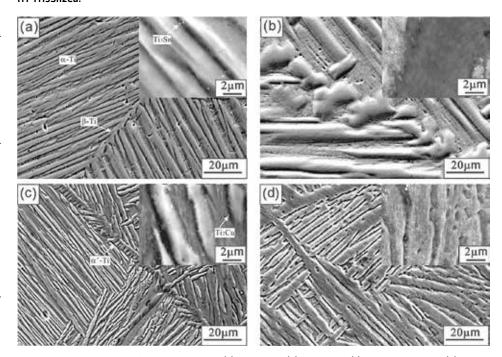


Figure 2: SEM micrograph of Ti15SnxCu alloys: (a) AS Ti15Sn, (b) HT Ti15Sn, (c) AS Ti15Sn2Cu and (d) HT Ti15Sn2Cu. The inserts show the high magnification microstructure.

structure of supersaturated Ti solid solution (α '-Ti), acicular α -Ti, and the pre-existing β -Ti grains. Hsu et al. [11] reported that the Ti5Sn alloy is comprised mainly of α '-Ti phase [34]. Others have reported that α '-Ti plates nucleate and grow from the prior β -grain boundaries at high cooling rates [12,35]. The data reported here is in good agreement with the results from the previous studies [23]. The SEM image of AS Ti15Sn alloy shown in Figure 2a displays the same structure by OM. It is found that the lamellae structure of heat-treated (HT) Ti15Sn alloy was coarser than that of AS Ti15Sn alloy, as shown in Figures 1b and 2b. From OM (Figures 1c, 1d) and SEM (Figures 2c, 2d), we can see that the 2 wt. % Cu content slightly influenced the microstructure of the alloys.

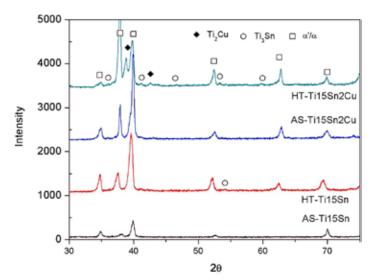


Figure 3: XRD patterns of the Ti15SnxCu alloys^{23,36}.

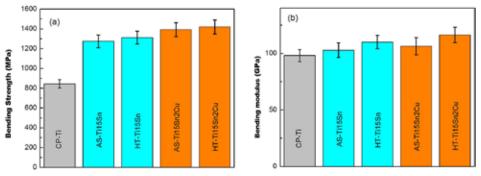


Figure 4: Bending property of Ti15SnxCu under different treatment condition: (a) bending strength and (b) bending modulus.

Alloy	Alloying element (wt%)	Bending strength (MPa)	Bending modulus (GPa)	Ref.
СР-Ті	-	843.2 ± 42.2	98.1 ± 5.3	
ATi15Sn	-	1275.3 ± 63.2	102.9 ± 6.5	
HT Ti15Sn	-	1312.7 ± 65.4	110.1 ± 5.9	
AS Ti15SnxCu	2Cu	1392.2 ± 69.5	106.4 ± 7.6	
HT Ti15SnxCu	2Cu	1420.3 ± 71.1	116.4 ± 6.8	
Ti5SnxCr	0 ~ 13Cr	1657 - 2505	97 -153	34
Ti6Al7Nb	-	1943 ± 26	111 ± 3	45
Ti15Mo1Bi	-	1975 ± 38	90 ± 1	
TixSi 1	~ 12.5Si	1621 - 2231	120 - 180	46
TixMo	7.5 ~ 20Mo	1395 -1650	55 - 87	47

Table 1: Bending property of Ti15SnxCu under different treatment condition.

In comparison to Ti15Sn alloy, the lamellar structure of AS Ti15Sn2Cu alloy (Figure 2c) was smaller than that of AS Ti15Sn alloy (Figure 2a). Likewise, the lamellar structure of HTTi15Sn2Cu alloy (Figure 2d) was smaller than that of HTTi15Sn alloy (Figure 2b).

The inserted high magnification SEM microstructure shows there is a small amount of nanoparticles distributed on the matrix. The EDS results indicated mainly consisted of Ti and Sn elements with an atom ratio of Ti to Sn of about three, corresponding to nano-Ti $_3$ Sn phase in the XRD pattern in Figure 3 [23,36]. It can be seen that the AS Ti15Sn alloy consisted of supersaturated Ti solid solution (o'-Ti),

acicular α -Ti, and nano-Ti Sn. After heat treatment, Figure 2b, nano-Ti₃Sn phase was observed on the matrix. The XRD pattern of the HT Ti15Sn alloy exhibited α'/α -Ti and Ti₃Sn peaks. According to the Ti-Sn phase diagram [37], the α + Ti Sn formed from the supersaturated α' -Ti phase due to the decline of Sn solubility in α -Ti phase as the temperature decreased from 881°C to room temperature; i.e.,

$$\alpha'-Ti \rightarrow \alpha + Ti_3Sn$$
 Equation 3

As indicated in an early report of Bulanova et al. [38], the Ti3Sn phase precipitated from the primary $\alpha\text{-Ti}$ phase in the Ti15Sn5Si alloy on cooling. Park et al. [39] reported that the $\text{Ti}_{82}\text{Sn}_{18}$ eutectic alloys present a mixture of the hexagonal $\alpha\text{-Ti}$ (P63/mmc) solid solution phase and the hexagonal Ti_3Sn (P63/mmc) intermetallic compounds (IMCs) [39,40]. Therefore, it is comfirmed that the nano-Ti $_3\text{Sn}$ phases were precipitated on the lath site [23].

The inserted high magnification SEM image of Ti15Sn2Cu alloy shown in Figures 2c and 2d displays many particles distributed on the matrix. The XRD pattern of the AS Ti15Sn2Cu alloy exhibited some

diffraction peaks, which indicated α'/α -Ti and Ti₂Cu phases. After heat treatment, the Ti15Sn2Cu alloy exhibited α'/α -Ti and nano-Ti₂Cu phases (Figure 2d). According to the Ti-Cu phase diagram [19] and many reports [10,41-43], the final equilibrium microstructure of the Ti-Cu alloy with lower Cu content consists of α -Ti and Ti₂Cu, which is formed by a eutectoid reaction. The body centered-tetragonal (bct) Ti₂Cu phase (MoSi2 type, a = 0.2943 nm, and c = 1.0786 nm) should be present in almost all the Cu-bearing titanium alloys [44]. Yao et al. [22] reported that the spherical Ti2Cu precipitates were observed to nucleate at grain boundaries and

intra-grains. Therefore, the strengthening mechanisms can be attributed to the dislocations around the Ti2Cu particles. This mechanism could explain how the Ti2Cu intermetallic restricts the grain growth in the Ti15Sn2Cu alloy, leading to the smaller grain size of Ti15Sn2Cu alloy than of Ti15Sn alloy. It was expected that Ti15Sn2Cu alloy would have better mechanical behavior than Ti15Sn alloy.

3.2 Bending strength

The bending strengths and bending moduli of the Ti15SnxCu samples are shown in Table 1. All the Ti15SnxCu alloys had significantly higher bending strengths (1,275.3 \pm 63.2 - 1,420.3 \pm 71.1 MPa) than the CP-Ti (843.2 \pm 42.2 MPa). It is obvious the bending strength of Ti15SnxCu was about 1.5 times that of CP-Ti. Numerous studies have shown the bending strength of new Ti alloys is about 1,395-2,505 MPa; i.e., 1,657-2,505 MPa for Ti5SnxCr (x = 0 - 13 wt. %) [34], 1,943 \pm 26 MPa for Ti6Al7Nb, 1,975 \pm 38 MPa for Ti15Mo1Bi [45], 1,621-2,231MPa for TixSi (x = 1 - 12.5 wt.%) [46], and 1,395-1,650 MPa for TixMo (x = 0 - 13 wt. %) [47]. The bending strengths of the Ti15SnxCu alloys were similar to that of TixMo Alloy.

The bending strength resulting from the AS condition was slightly lower than that resulting from the HT condition: 1,275.3 \pm 63.2 MPa vs. 1,312.7 \pm 65.4 MPa for Ti15Sn, and 1,392.2 \pm 69.5 MPa vs. 1,420.3 \pm 71.1 MPa for Ti15Sn2Cu (Table 1 and Figure 4a), respectively. Comparing the different conditions (AS vs. HT), the relative bending strengths of both Ti15Sn and Ti15Sn2Cu sample were respectively 2.9% and 2.0% higher under the HT condition than under the AS condition (Table 2 and Figure 5). Comparing OM, SEM and XRD revealed that the bending strength of the HT sample increased, obviously due to the formation of the coarser α'/α -Ti phase.

In Ref [48], the bending strength is proof of actual α -Ti phase precipitation. Also, it can be seen from Table 3 and Figure 5 that the bending strength values of both AS and HT Ti15Sn2Cu samples were higher than that of both AS and HT Ti15Sn sample (about 9.2% and 8.2%, respectively).

The addition of 2 wt.% Cu caused a large number of nano- $\mathrm{Ti}_2\mathrm{Cu}$ particles to precipitate in the martensite structure, and the nano- $\mathrm{Ti}_2\mathrm{Cu}$ particles enhanced the bending strength of the Ti15Sn2Cu alloy by increasing the resistance to obstacle dislocation movement. In previous work, both $\mathrm{Ti}_2\mathrm{Cu}$ and $\mathrm{Ti}_3\mathrm{Sn}$ nanoparticles was confirmed by transmission electron microscopy (TEM) [12]. From the Ti-Cu phase diagram, it is clear the nano- $\mathrm{Ti}_2\mathrm{Cu}$ particles precipitated in the eutectoid reaction (at 790°C) from β phase [19,22],

β -	$\rightarrow \alpha$	$+ Ti_2$	Cu	Equation 4
ρ -	$\rightarrow u$	T 112	Cu	Equation 4

The Ti_2Cu particles mainly precipitated from the supersaturated α '-Ti phase due to the decline of Cu solubility in α -Ti phase as the temperature decreased from 790°C to room temperature; i.e.,

$$\alpha' \rightarrow \alpha + Ti_2Cu$$
 Equation 5

The XRD results in Figure 3 indicate the intensity of the Ti_2Cu peak (2θ = 38.33) increased after heat treatment, leading to a high volume fraction of Ti_2Cu . Thus, the bending strength of the HT Ti15Sn2Cu alloy was higher than those of the as-cast ones. Kikuchi et al. [18] noted that Ti_2Cu precipitates can strengthen Ti alloy. Sun et al. [21] reported nano- Ti_2Cu precipitates hinder the dislocation movement and thereby increase the mechanical property. A number of articles have shown Ti_1Cu and strengthen Ti_1Cu alloys due to the solid-solution strengthening effect and the formation of Ti_2Sn in the alloy [29,30].

The Bending modulus results are shown in Table 1 and Figure 4b. The bending moduli $(102.9 \pm 6.5 \text{ GPa} - 116.4 \pm 6.8 \text{ GPa})$ of all the Ti15SnxCu alloys were higher than that of CP-Ti (98.1 ± 5.3 GPa). The bending moduli of the Ti15SnxCu alloys were similar to those of Ti6Al7Nb (111 ± 3 GPa) [45] and Ti5SnxCr alloy (97-153 GPa) [34]. To simplify the analysis (Table 2 and Figure 5a), the relative ratios of the bending moduli were higher for the HT condition (+ 7.0% for Ti15Sn alloy and + 9.4% for Ti15Sn2Cu alloy) than for the AS condition. Relative to the Ti15Sn alloy (Table 3 and Figure 5b), the bending modulus of the Ti15Sn2Cu alloy was enhanced by 3.4% in the AS condition and by 5.7% in the HT condition. A higher elastic modulus is

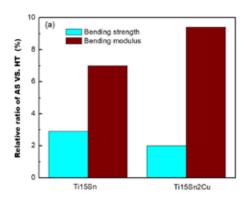
essential to minimize elastic deformation. In addition, Kikuchi et al. [49,50] and Guillet et al. [51] reported hard IMCs generally have a higher elastic modulus than the constituting elements and the increase in the modulus of Ti15Sn2Cu alloy was due to the formation of the Ti2Cu phase in the α matrices. The elastic modulus is one of the most important properties of a dental structural alloy, such as bridges and partial denture frameworks. In general, a higher elastic modulus is also beneficial for machining titanium alloys. Therefore, the HT

Alloy	Bending strength	Bending modulus
Ti15Sn	+ 2.9	+ 7.0
Ti15Sn2Cu	+ 2.0	+ 9.4

Table 2: The relative ratios (%) of bending strength and bending modulus of AS versus HT of Ti15SnxCu.

Condition	Bending strength	Bending modulus
AS	+ 9.2	+ 3.4
HT	+ 8.2	+ 5.7

Table 3: The relative ratios (%) of bending strength and bending modulus of Ti15Sn versus Ti15Sn2Cu alloy.



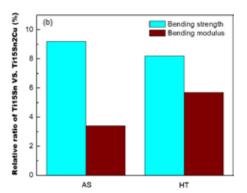
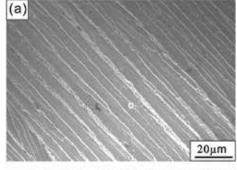
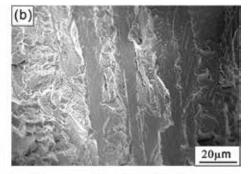
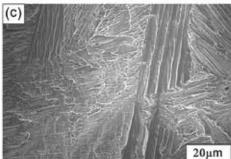


Figure 5: The relative ratios (%) of both bending strength and bending modulus: (a) AS versus HT and (b) Ti15Sn versus Ti15Sn2Cu.







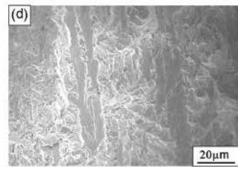


Figure 6: SEM fractographs of Ti15SnxCu alloys: (a) AS Ti15Sn, (b) HT Ti15Sn, (c) AS Ti15Sn2Cu and (d) HT Ti15Sn2Cu.

Ti15Sn2Cu alloys possessed better bending strengths and moduli mainly due to the strengthening effects of the crystallites with fine grain sizes and nano-Ti₂Cu precipitates.

3.3 Fracture surface

Figures 6 and 7 show SEM micrographs of the fractured surfaces of the Ti15SnxCu specimens after three-point bending tests. Figure 6a presents two common fracture surfaces of AS Ti15Sn alloy after three-

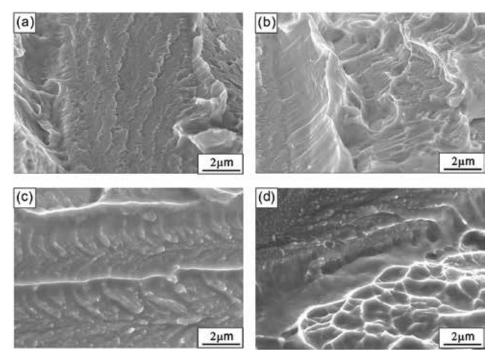


Figure 7: The higher magnification SEM fractographs of Ti15SnxCu alloys: (a) AS Ti15Sn, (b) HT Ti15Sn, (c) AS Ti15Sn2Cu and (d) HT Ti15Sn2Cu.

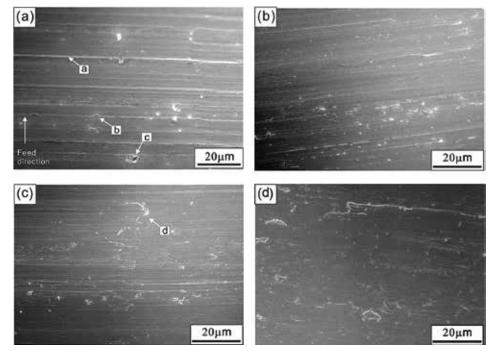


Figure 8: Cut surfaces of Ti15SnxCu samples: (a) AS Ti15Sn, (b) HT Ti15Sn, (c) AS Ti15Sn2Cu and (d) HT Ti15Sn2Cu.

point bending tests. The surfaces were mostly interlamellar fractured (gray smooth) and a few, narrow translamellar fractured (light rough). Interlamellar ruptures were clearly visible in the fractured structures of the AS Ti15Sn alloy at the higher magnification of SEM fractographs (Figure 7a). The river pattern was clearly visible. After heat treating, the fracture surface in Figure 6b appeared to be mixed interlamellar/translamellar. In the high-magnification image in Figure 7b, the dimples can be seen clearly. Since interlamellar strength is significantly lower than translamellar strength, the strength of the HT Ti15Sn sample was higher than that of the AS Ti15Sn sample.

The effect of adding 2 wt% Cu into the Ti15SnxCu alloy is visible in the fracture images in Figure 6c and 6d. The randomly oriented

colonies (with respect to the loading direction) and mostly translamellar fracture were observed. A high magnification image of the interlamellar area is given in Figure 7c, where white nanoparticles can be seen clearly. According to energy dispersive spectroscopy (EDS) analysis, the chemical composition (wt. %) of the white nanoparticles was Ti (83.98%), Sn (3.82%), and Cu (12.20%), which approximates that of the nano-Ti₂Cu phase. The same phenomenon was also observed in the heat treated Ti15Sn2Cu sample (Figures 6d and 7d), confirming the nano-Ti₂Cu particles could improve the machine property of the Ti15Sn2Cu samples. It is clear that, due to the higher bending strength of Ti15Sn2Cu alloy, the brittle nano-Ti₂Cu phase precipitated, and those precipitates restricted the motion of dislocations in the multiple slip systems and enhanced the bending strength. It can also be speculated that the cracks initiated and propagated in the brittle nano-Ti₂Cu particles in an early stage of the shearing.

3.4 Machinability

Figure 8 shows the cut surfaces of the Ti15SnxCu alloys. Cutting scars (mark a) are obvious on the cutting surface of the AS Ti15Sn alloy (Figure 8a), as well as small amounts of smears (mark b) and microvoids (mark c). The cutting surface of the HT Ti15Sn2Cu alloy (Figure 8d) had fewer cutting marks than did those of the other samples.

The order of cutting traces in the machinability testing was as follows: AS Ti15Sn > HT Ti15Sn > AS Ti15Sn2Cu > HT Ti15Sn2Cu.

Compared with that of Ti15Sn alloy, the cutting surface of Ti15SnxCu alloy with 2 wt. % Cu had increased tearing (mark d). Lower ductility is generally beneficial to the machinability of metals [52]. It is reported that scar morphologies are related to grind adhesion, and the relative low hardness and strength of alloys cause chatter vibration or deformation of the work piece during machining [11,53]. Figure 9 shows SEM images of the metal chips resulting from the machinability testing of the Ti15SnxCu. The metal chips of the Ti15Sn2Cu alloy were finer than those of the Ti15Sn alloy. As shown in Figure 10, the aver-

age size of the metal chips of Ti15Sn2Cu alloys in order is: AS Ti15Sn (645 \pm 56 $\mu m)$ > HT Ti15Sn (547 \pm 42 $\mu m)$ > AS Ti15Sn2Cu (354 \pm 46 $\mu m)$ > HT Ti15Sn2Cu (286 \pm 32 $\mu m)$. The HT Ti15Sn2Cu alloy produced much finer metal chips, which are reported to be more suitable for machinability [20,54]. From these experiments, it is obvious the nano-Ti2Cu precipitates caused a loss of ductility and facilitated more effective machinability of Ti15Sn2Cu alloy because small broken fragments more readily formed [20].

4 CONCLUSION

In the current investigation, the effect of heat treatments on the microstructure, mechanical, and machinability properties of

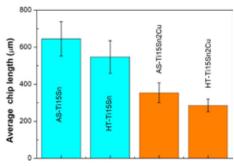


Figure 10: Average length of metal chips of Ti15SnxCu alloys.

Ti15SnxCu was investigated, and the following points were concluded:

- **1.** The Ti15Sn alloy exhibits (fine α'/α -Ti + nano-Ti₃Sn) and (coarse α'/α -Ti + Ti₃Sn) phases under the as-cast and HT conditions, respectively. However, Ti15SnxCu with 2 wt. % Cu (Ti15Sn2Cu) alloy comprises α'/α -Ti + nano-Ti₂Cu phase under such conditions.
- **2.** All the Ti15SnxCu alloys had higher bending strengths $(1,275.3 \pm 63.2 1,420.3 \pm 71.1 \text{ MPa})$ than that of CP-Ti $(843.2 \pm 42.2 \text{ MPa})$. Notably, the bending strengths of the HT alloys were higher than those of theAS

alloys. The HTTi15Sn2Cu alloy also had the highest bending strength and modulus, mainly due to the strengthening effects of the crystallites with fine grain sizes and nano-Ti Cu precipitates.

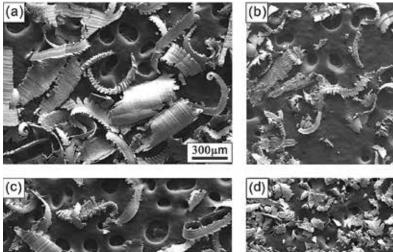
- **3.** In the AS T15Sn alloy, the predominantly observed fracture mode was interlamellar-fracture. In the HTTi15Sn alloy, the fracture mode was mixed interlamellar/translamellar. However, adding 2 wt. % Cu to the Ti15SnxCu alloy change the fracture structures to mostly translamellar fracture, especially in the case of the heat treated Ti15Sn2Cu alloy.
- **4.** It was confirmed the nano-Ti₂Cu precipitation was caused by crack initiation and propagation, which reduced ductility and improved the machinability of Ti15Sn2Cu alloy.

5 ACKNOWLEDGMENTS

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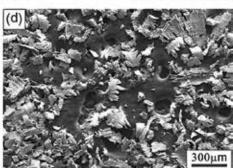


Figure 9: Metal chips cut from Ti15SnxCu samples: (a) AS Ti15Sn, (b) HT Ti15Sn, (c) AS Ti15Sn2Cu and (d) HT Ti15Sn2Cu.

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